Chemometrical approaches for the characterization of the environmental behavior of fragrances



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Abstract

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Fragrances represent a new class of emerging pollutants. These compounds, which are used in many consumer products, including cleaning and washing agents, and personal care products, are often released directly to the environment, posing a risk for the exposed organisms. Concerns from the environmental presence of these compounds is a result of their potential persistence and/or toxicity in environmental media and of their possible effects on humans (asthma, allergies, headaches...), as well as of their ability to accumulate in adipose tissue such as breast milk. Moreover, since the fragrances are often volatile compounds, they are connected to both indoor and outdoor air pollution.

Unfortunately little is known about the environmental occurrence and fate of these substances which potential effects on humans and aquatic ecosystems are not yet clearly understood. The use of predictive approaches based on chemometrical techniques, such as Quantitative Structure Activity (Property) Relationships (QSA(P)R), can help in filling this data gap and characterizing the environmental and ecotoxicological profile of these substances.

In this study different Ordinary Least Squares (OLS) regression-based QSA(P)R models were developed for toxicological and physico chemical endpoints. Theoretical molecular descriptors were calculated by DRAGON software, and the best modeling variables were selected also by applying Genetic Algorithms (GA).

The developed models could be particularly useful for characterization, screening and prioritization of widely used fragrances compounds, and also a priori, in the design of new products as safer alternative to the existing dangerous.

Materials and Methods

Data Set: The experimental data related to toxicological and physico-chemical data were taken partially from literature MULTIPLE LINEAR REGRESSION MODELS and Variable Selection were performed by [1,2,3,4,5,6] and partially from available online databases [7,8,9]. Ordinary Least Squares regression (OLS) and Genetic Algorithms (GA) method [12]. Toxicological endpoints modeled are LD50 Oral Mouse [1,2,7], inhibition of NADH oxidase (EC50 NADH-Ox) and the EXTERNAL VALIDATION effect on mitochondrial membrane potential (EC50 ΔΨm)[3]. Prediction set selection was based on the molecular structure (by Kohonen Maps - Artificial All the responses have been transformed to logarithmic units and, if necessary, multiplied by -1 to obtain positive Neural Networks (K-ANN) or using the Random by response approach. values. Physico-chemical properties modeled are LogKow, Water Solubility (WS) and Vapor Pressure (Vp); like for TOOLS OF VALIDATION AND DIAGNOSTICS toxicological endpoints, the responses have been transformed to logarithmic units. Molecular Descriptors: Models were developed taking into account the recently proposed OECD principles for QSAR 452 molecular descriptors (OD,1D,2D,3D) were calculated for Physico-chemical properties and LD50 Oral Mouse and validation [13]. 459 for the others toxicological endpoints by the software DRAGON [10].

4 quantum-chemical descriptors [Highest Occupied Molecular Orbital (HOMO), Lowest Unoccupied Molecular Orbital (LUMO), HOMO-LUMO gap (DHL) and the ionisation potential (P ion)] were calculated. The input files for descriptor calculation were obtained by the Semi empirical (AM1) Hamiltonian for the geometry optimisation method available in the HYPERCHEM package [11].

Internal (by Q²_{LOO} and Q²_{boot}, Y-scrambling) and external validation (verified by Q²ext) [14].

R²

89

· Check of the quality of the best models by Residuals and Williams plot

· Applicability Domain verified by leverage approach.

Results Q²BOOT

 $\mathbf{Q}^2_{\mathrm{LOO}}$

86.19 81

91.35 86.15 80.47

91.69 88.55 83.94

 90.29
 84.31
 74.01

 90.88
 85.8
 78.78



Principal Component Analysis (PCA) was performed on experimental and reliable predicted values of LogKow, LogWs and LogVp available for 53 compounds. This analysis was applied in order to have a multivariate view and a graphic representation of the potential overall environmental partitioning profile of the studied fragrances.

Chemicals are ranked along PC1 and PC2 (Tot E.V.%=93.9; PC1 E.V%=76.6%), according to their physico-chemical properties and (from right to left) according to increasing molecular weight (MW). Hydrophobic compounds, such as Musks, are located on the left side of the graph (zone 1 accumulation potential in biota, soil and sediments); soluble and volatile fragrances are ranked on the right of PC1 and further along PC2. In particular volatile fragrances, such as linalool derivatives, are placed in zone 2; more soluble fragrances, such as low-MW salicilates and cinnamates are placed in zone 3.

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RMSE

Train

0.265 0.226

Test

0.061

0.07

0.23

0.22

 $\mathbf{Q}^2_{\mathbf{EXT}}$

73.4

97.26 0.19

98.19 0.2

88.78 83.41 72.2 90.23 0.256 0.186

R²-YSc

9.25

14.34

16.88



Conclusions

1. Limited availability of experimental data utilizable for QSAR (in particular for SIDS and toxicological endpoints).

2. New QSAR and QSPR models have been developed for the prediction of some physico-chemical (LogKow, LogWS, LogVp) and toxicological endpoints (acute oral mouse toxicity, and two endpoints related to mitochondrial toxicity [Inhibition of NADH-Oxidase and the effect on mitochondrial membrane potential (ΔΨm)).

Despite the limited amount of available data, all the models were carefully internally and externally validated. At our knowledge no other QSAR/QSPR models are available in literature for these endpoints.

3. All the variables selected in the proposed models are 2D descriptors, independent of chemical conformation. Quantum chemical descriptors as well as LogP were not selected in the externally predictive models.

4. The combination of the predicted and experimental physico-chemical properties data by PCA allowed for the identification of a profile of the environmental behavior of fragrances. Three zones were identified which distinguish among volatile (linalool derivatives), soluble (salicilates and cinnamates) or adsorbed (musks) compounds.

New data on the real persistence of these compounds could help in the refinement of the hazard profile of these fragrances.